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It was found: (1) that in increasing the water-vapor
tension over heated orthophosphoric acid the tempera-
ture of formation of the pyro- acid rises faster than
that of the meta- acid. (2) On account of the volatility
of the meta- acid the investigation could be carried only
to a relatively low temperature so that the anticipated
meeting of the curves of temperature formation for
pyro- and meta- acid, that is, the direct transition of
ortho- into meta- acid, could not be reached. (Author)

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